

84957

Z/003/60/054/011/003/005

E112/E453

Vapour-Phase Partition Chromatography of Mixtures of Hydrogen, Paraffins and Olefins, Using the Micro-Flare Detector

olefins. Modifications are: Replacement of nitrogen as carrier gas by air. It is claimed to lead to a more perfect combustion of the eluted components and to greater constancy of the length of flame. Lengthening of the flame during elution is held to be responsible for some inaccuracies in the Wirth method. It is also held that the temperature of the flame contributes more to the sensitivity of the method than its length. A schematic arrangement of the chromatographic apparatus is given: Hydrogen, from a cylinder, and air, from a compressor, are reduced to constant pressure through diaphragm valves. The flow of the gases is controlled by a second set of valves and metered through a capillary flowmeter and through drying tubes, filled with magnesium chloride to the heads of the columns. Air passes through a chromatographic column packed with activated sodium-aluminium silicate (Alusil) and provided with a sample introducing device. The column is placed in an electrically heated furnace, the temperature of which is raised during analysis from 20 to 190°C. Hydrogen passes through a dummy column and is mixed with the carrier gas immediately before

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combustion. Details of detector construction are given. Several variables in flare design were considered. Best results were obtained with a platinum capillary with a bore of 0.20 to 0.28 mm. (Wirth uses stainless steel.) The effects of ratios of hydrogen to carrier gas on the sensitivity of the method was studied. Higher hydrogen contents in the gas mixtures give higher base lines and more complete combustion. Lower base lines, however, show better peak characteristics. The author has established experimentally that best results were obtained with a hydrogen flow rate of 0.45 ml/sec for an air flow of 0.35 ml/sec. Under these conditions, the basis temperature of the flame, determined by the thermocouple at a distance of 8 mm from the tip of the flame, amounted to 400°C. Different types of thermocouples were considered. Iron-constantan was found the most promising. However, owing to rapid oxidation of the iron wire, variations in the constants of the thermocouple were experienced. Attempts to improve the thermocouple stability are described. Sealing into a thin-walled glass capillary failed, owing to

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excessive thermal inertia. Similar, not quite satisfactory results were obtained with copper-constantan. Best results were finally achieved by applying a silicone protective coating to iron constantan. Response to different types of thermocouples was tested by following the partition chromatogram of hydrogen-methane mixtures and results are presented graphically. The superiority of silicone-coated iron-constantan is clearly demonstrated. Results of the chromatographic partition of hydrogen, paraffins and olefins from C₁ to C₆ - the object of the present study - are tabulated, using the micro-flare detection and comparing to detection by the gas density balance method. A very good agreement between the results by both methods was established. The method permits the determination of hydrogen in the presence of hydrocarbon mixtures. No oxidation of the hydrocarbons by the carrier gas (air) was noticed even when the temperature of the chromatographic column was raised to 190°C. Results are based on 10 determinations and showed good reproducibility. Acknowledgments to Engineer J. Janák, Gas Analysis Laboratory, ČSAV, Brno, for supplying data

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on the design of flame detector. There are 5 figures (showing
a schematic arrangement of the chromatographic apparatus, a drawing
of the cross section of the detector, the electrical circuit
diagram and 2 graphs), 2 tables and 5 references. 2 Czech and
3 English.

ASSOCIATION: *Stalinovy závody, n.p., Záluží v Krusných horách*
(Stalin Works, Záluží)

SUBMITTED: November 17, 1959

✓

Card 5/5

BARENBOYM, A.M., kand. tekhn. nauk; GALIYEVA, T.M., inzh.;
GINZBURG, D.B., prof.; GRISSIK, A.M., inzh.; ZIMIN, V.N.,
dots.; KUSYAK, V.A., kand. tekhn. nauk; RUTMAN, E.M.,
inzh.; KHODOROV, Ye.I., kand. tekhn. nauk; CHIERSKIY,
A.F., kand. tekhn. nauk

[Heat calculations for furnaces and dryers of the silicates
industry] Teplovye raschety pechei i ushilok silikatnoi
promyshlennosti. Izd.2., perer. i dop. Moskva, Stroiz-
dat, 1964. 495 p. (MIRA 17:12)

AKT FAKULTET

ORAZMETOV, Z.; GORELKIN, L.M.; POTYAYEV, M.Ye.; ZARUDI, Ye.O., metodist;
MITELEV, V.S.; VASIL'YEV, A.V.; GORSHENKOV, N.G.;
RUTKOVSKIY, O.O.; KUSYAPKULOVA, T.Sh.

Letters to the editors. Geogr. v shkole 22 no.2:72-76
Mr-Ap '59. (MIRA 12:6)

1. 1-ya shkola pos. Andreyevka Turkmenskoy SSR (for Orazmetov).
2. Shkola pri shakhte No.11 Karachayevskogo rayona Stavropol'-skogo kraya (for Gorelkin).
3. Andreyevskaya semiletnyaya shkola Penzenskoy oblasti (for Potyayev).
4. Bashkirskiy institut usovershenstvovaniya uchiteley (for Zarudi).
5. Rayonnyy pedagogicheskiy kabinet s.Kich-Gorodok Vologodskoy oblasti (for Mitelev).
6. Alekseyevskaya shkola Stalingradskoy oblasti (for Vasil'yev).
7. Takhromskaya shkola No.2 Moskovskoy oblasti (for Goshenkov).
8. 4-ya shkola g.Alma-Ata (for Rutkovskiy).
9. 64-ya shkola g.Alma-Ata (for Kusyapkulova).

(Geography--Study and teaching)

L 26505-66 EWP(m)/EWT(1) GS

ACC NR: AT6008147

UR/0000/65/000/000/0072/0080 36

B+1

AUTHOR: Saykovskiy, M.I.; Dorfman, A.Sh. (Candidate of technical sciences); Didenko,
O.I.; Kusyuk, A.I.; Stepanenko, A.P.

ORG: None

TITLE: Results of aerodynamic investigation of the compressor intake on models and
in full scale

SOURCE: AN UkrSSR. *Techeniya zhidkostey i gazov* (Flows of liquids and gases) Kiev,
Naukova dumka, 1965, 72-80

TOPIC TAGS: compressor design, aerodynamic test, test model

ABSTRACT: The paper describes scale model and full scale aerodynamic tests on compressor intakes. Rigidly oriented 3-channel total + pressure tubes installed in a rotatable ring were used to measure the flow turning angle, velocity, and total air pressure. Schematics of the compressor intake are shown. The energy loss coefficient, ξ , of the intake was calculated from the average loss of total pressure, Δ_o , the average ram density, ρ , the average normal velocity, v_n , and the compressibility correction factor δ ($\delta = 1 - M^2/4$) using: $\xi = 2\Delta_o/\rho_0 v_n^2$. (1) Conditions and measurement results are given for 12 design variants. All variants show a fairly uniform distribution of velocities over the cross sections. Losses are comparatively low in all variants, somewhat

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ACC NR: AT6008147

lower for the design with a diagonally disposed entrance. Hints for efficient compressor intake design are discussed, among them the necessity to have adequate overall axial dimensions so as not to increase unduly the curvature at flow bends. Model tests have indicated a sufficiently close correspondence of the flow rotation angles and velocity distributions with the full scale data. Orig. art. has: 4 figures, 1 formula.

SUB CODE: 13/ SUBM DATE: 01Sep64

Card 2/2 CC

KUSTURINA, L.A.

Producing mutants of mold fungi by the use of ultraviolet rays.
Report No2: Morphological characteristics of variants of *Aspergillus*
nidulans produced through irradiation with various doses of ultra-
violet rays. *Mikrobiologija* 28 no.5:667-674 S-0 '59. (MIRA 13:2)

1. Institut mikrobiologii AN SSSR.
(ASPERGILLUS radiation eff.)
(ULTRAVIOLET RAYS eff.)

L 22372-(6) EMP(e) IJP(c) AI/WH
ACC NR: AP6009605

SOURCE CODE: FO/0045/65/028/004/0491/0497

49

B

AUTHOR: Sujak, B.; Kusz, J.

ORG: Laboratory for Induced Electron Emission, Institute of Experimental Physics, Wroclaw University; Department of Experimental Physics, Pedagogical College of Opole

21. 44. 85

TITLE: Field-excited (exo)electron emission from Rochelle salt crystals

SOURCE: Acta physica polonica, v. 28, no. 4, 1965, 491-497

TOPIC TAGS: electron emission, barium titanate, Rochelle salt, electron polarization, surface ionization

ABSTRACT: (Exo)electron emission excited by an externally applied electric field was found to occur from crystalline Rochelle salt specimens and from ceramic samples of the type of barium titanate. Subsequent to polarization in a sufficiently intense electric field, a Rochelle salt specimen becomes a source of exoelectrons without requiring light stimulation. The lowest polarizing voltage U_p at which exoelectrons begin to be recorded definitely depends on the crystallographic plane parallel to which the specimen has been cut. U_p is lowest in specimens cut parallel to the plane (100). Its value, as

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ACC NR: AP6009605

well as the variations in the recorded intensity of exoelectron emission, depend markedly on the previous history of the specimen. The results make it highly plausible that this "emission" is due to surface ionization of the gas wherein the sample is immersed. Orig. art. has 5 figures and 3 formulas. [Based on author's abstract] [KS]

SUB CODE: 20/ SUBM DATE: 09Mar65/ ORIG REF: 002/
SOV REF: 002/ OTH REF: 001

Card 2/2ddr

ELEK, Janos, akadémiai doktor; HALL, János, akadémiai doktor;
György; KOMORNÍK, Lajos; KOMORNÍK, György; KOMORNÍK, Lajos;
Jene, Pál; KOMORNÍK, Vilmos

Efficiency of gas consumption of various consumers with special
regard to the optimum utilization form energy carriers. Preprint
es atom 16 no.1:15-17 Ja '65.

1. National Petroleum and Gas Industry Plant, Budapest (see
Porty).
2. Ministry of Heavy Industry, Budapest (see
Porty).

100
S/644/62/009/006/049/127
B156/B112

14 6510
24 5242

AUTHORS: Zelazny, R., Kuzele, A., Mika, J.

TITLE: Solution of the one-velocity Boltzmann equation with first-order anisotropic scattering of neutrons in plane geometry

PERIODICAL: Referativnyy zhurnal. Matematika, no. 6, 1962, 96, abstract 68406 ([referat] Inst. badań jadrow. PAN, v. 9, no. 216, 1961)

ABSTRACT: The Keys method of solving one-velocity transfer equations is applied to the case of anisotropic scattering. The case of a linear scattering indicatrix is examined in detail. The solution is sought by separating the variables. To find the unknown functions, the parameters of the solution, a system of two singular integral equations is set up. Systems of orthogonal functions of an angular variable are studied in detail and used for solving the problem. In conclusion, an example of the solution to the problem of finding the reflection factor of a semispace is given. It is stated in this article that the algorithm developed here can also be applied to the case of a more complex scattering indicatrix.
[Abstractor's note: Complete translation.]

Card 1/1

POLACZEK, Lucyna, mgr; KUSCZAK, Halina; FISCHHOFF, Kazimiera

Method of determining ethyl flavono-7-hydroxyacetate and some possible impurities from its synthesis. Chem anal 9 no.2: 275-281 '64.

1. Zaklad Analityczny, Instytut Farmaceutyczny, Warszawa.

KUSZELEWSKI, Leszek

Studies on the use of manure. Pt. 3. Rocznik nauk rolniczych
87 no.2:251-287 '63.

1. Katedra Chemii Rolniczej, Szkoła Główna Gospodarstwa
Wiejskiego, Warszawa.

KUSZELEWSKI, Leszek

Studies on manuring. Pt. 4. Rocznik nauk rolniczych 86 no. 4: 543-
572 '62.

1. Katedra Chemii Rolniczej, Szkoła Główna Gospodarstwa
Wiejskiego, Warszawa.

KUSZELEWSKI, Leszek

Studies on manure application. Pt.2. The influence of the methods
of application of manure in the cultivation of potatoes. Rocznik nauk
roln rosl 81 no.3:577-619 '60. (EEAI 9:10)

1. Zaklad Chemii Rolniczej Szkoły Głównej Gospodarstwa Wiejskiego
w Warszawie. Kierownik: Prof. dr. M.Gorski.
(Poland--Fertilizers and manures)
(Poland--Potatoes)

KUSZELEWSKI, Leszek; PENTKOWSKI, Andrzej

Properties of manure submitted to methane fermentation in the light of
pot tests. Rocznik nauk roln. 82 no. 3: 715-737 '61.

1. Katedra Chemii Rolniczej, Szkoła Główna Gospodarstwa Wiejskiego,
Warszawa i Pracownia Metodyczno-Nawozowa, Instytut Uprawy, Nawożenia i
Gleboznawstwa, Warszawa; Kierownik: prof. dr. M. Gorski.

27319
P/046/60/005/011/008/018
D249/D303

24.1000

AUTHORS: Kuszell, Antoni, and Mika, Janusz

TITLE: Thermal utilization factor for a water-graphite
moderated lattice

PERIODICAL: Nukleonika, v. 5, no. 11, 1960, 743 - 754

TEXT: One- and two-group calculation methods of the thermal utilization factor for a heterogeneous reactor are discussed and compared. The standard two-group method is presented in a modified form by considering a non-zero neutron current at the boundary of the cell. The purpose of the work was to investigate a reactor design with a specified fuel element geometry. The problem was approached by introducing a simplified geometry of the fuel element and treating it as a homogeneous system using the Seitz-Wigner method. The one-group theory expression for the thermal utilization factor is shown for the case of cylindrical symmetry of the cell. It is pointed out that in order to avoid difficulties of evaluating the

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Thermal utilization factor ...

different thermal source densities of the fuel and of the moderator, one can assume the fast neutron flux as constant throughout the cell. Under this assumption it is possible to evaluate the ratio of the slowing-down powers of the fuel and moderator in terms of their known neutron scattering properties. The equations of the two-group theory are presented for cylindrical symmetry and a solution is given for a system with infinite number of identical cells. By assuming the multiplication constant K a variable parameter, the thermal utilization factor for a system of finite number of cells is obtained. It is stated, that the calculations considered so far imply an assumption of a zero neutron flux at the boundary of the cell. The authors present a calculation method based on assumption of a finite flux, and express the thermal factor in terms of a single-parameter. The latter is evaluated by means of the semi-empirical Gurevich-Pomeranchuk formula (Ref. 5: A.D. Galanin, Teoriya yadernykh reaktorov na teplovых neutronakh, Moskva (Theory of Thermal Neutron Nuclear Reactors, Moscow), 1957) and Deutsch's formula (Ref. 3: Reactor Physics Constants, ANL 5800). Finally the

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Thermal utilization factor ...

two-group theory is improved by including the age-diffusion corrections. Numerical calculations were performed for RFT fuel elements consisting of six concentric tubes of aluminum and uranium oxide (U^{235} 90 % enriched) ceramics cladded with aluminum, the whole surrounded by graphite, (Fig. 1) and water as coolant flowing through the space between the tubes. At the center of the element, there is an experimental channel filled with air. The results are presented in Table 2. In conclusion the authors remark that the remarkable consistency of the results obtained by different methods is implied by the characteristic properties of the system considered, and more significant deviations should be expected for a system with a more strongly absorbing moderator and with a smaller ratio of volumes of the fuel and the moderator, i.e. for smaller values of the thermal utilization coefficient f . There are 2 figures, 2 tables, and 6 references: 1 Soviet-bloc and 5 non-Soviet-bloc. The 4 most recent references to the English-language publications read as follows: A.C. Clark, D.A. Newmarch, AERE RP/R, 1657; Reactor Physics Constants. ANL 5800; R.L. Murray, Nuclear

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27319

P/046/60/005/011/008/018

D249/D303

Thermal utilization factor ...

Reactor Physics. Prentice Hall, 1957; H. Ritz, "Nucleonik" 1, no. 5, 175, 1959.

SUBMITTED: September, 1960

Fig. 1.

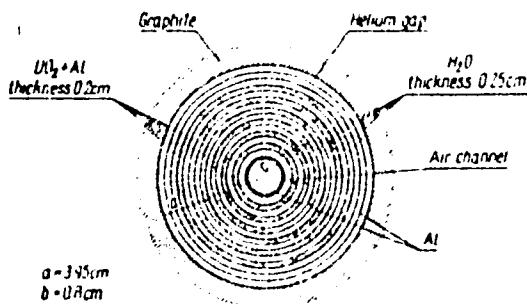


Fig. 1 Fuel element

Card 4/5

ZELAZNY, R.; KUSZELL, A.

Milne's problem for two adjacent half spaces. Bul Ac Pol mat
9 no.3:217-220 '61.

1. Institute of Nuclear Research, Polish Academy of Sciences,
Warsaw, and Institute of Theoretical Physics, University,
Warsaw. Presented by L. Infeld.

ZELAZNY, R.; KUSZELL, A.

Milne's problem for two adjacent half spaces. Bul Ac Pol mat 9
no.3:219-220 '61.

I. Institute for Nuclear Research, Polish Academy of Sciences and
Institute of Theoretical Physics, Warsaw University. Presented by
L. Infeld.

(Spaces, Generalized)

24.6510
S/044/62/000/006/048/127
B156/B112

AUTHORS: Zelazny, R., Kuszell, A.

TITLE: A special model of a two-group approach in neutron transport theory

PERIODICAL: Referativnyy zhurnal. Matematika, no. 6, 1962, 96, abstract 6B404 (bull. Acad. polon. sci. Sér. sci. math., astron. et phys., v. 9, no. 6, 1961, 461-466) fb

TEXT: A method is proposed for solving the equation for neutron transfer, which is confined to the two-group method. It is assumed that the free path lengths do not depend on the number of the group. A solution to the problem is sought using a Fourier transformation of a geometrical variable. The result is to reduce the problem to solving a system of integral equations for functions depending only on an angular component. When a system of eigenfunctions has been found, the solution to the problem is written in the form of a Neumann series. Various applications of the method are discussed, the most interesting of them being the results of solving the Milne problem in a two-group approximation and of solving the

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A special model of a two-group ...

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3156/B112

problem of the critical dimensions of a reactor in the form of an infinite layer. [Abstracter's note: Complete translation.]

VB.

Card 2/2

KUSZELL, A.

The critical problems for multilayer slab systems. Acta physica Pol
20 no.7:567-589 '61.

1. Institute of Nuclear Research, Polish Academy of Sciences, Warsaw.

KUSZEN, P.

①
Clinical aspects and pathology of poliradiculitis. P. Kuszen
(*Wiad med katol*, 1980, **103**, 201-204). A review with 5 case reports and 10 references.
W. R. Parr

BILOWICKA, Maria; LIBISZOWSKA-STANIUL, Maria; KUSZEWSKI, Bogdan;
SZELEZYNSKI, Kazimierz

Combined treatment of pulmonary tuberculosis with streptomycin
and isoniazid. Gruzlica 24 no.1:41-47 Jan 56.

1. Z Kliniki Ftyzjatrycznej Akademii Medycznej w Gdansku
Kierownik: prof. dr. M. Telatycki, Praca zlecona przez Instytut
Gruzlicy, Gdańsk, ul. Debinki 7 a. Klinika Ftyzjatryczna.
(TUBERCULOSIS, PULMONARY, ther.)

streptomycin & isoniazid.)
(STREPTOMYCIN, ther. use
tuberc., pulm., with isoniazid.)
(NICOTINIC ACID ISOMERS, ther. use
isoniazid in pulm. tuberc., with streptomycin.)

KUZMAK, Bogdan; KALIVAK, Iwan

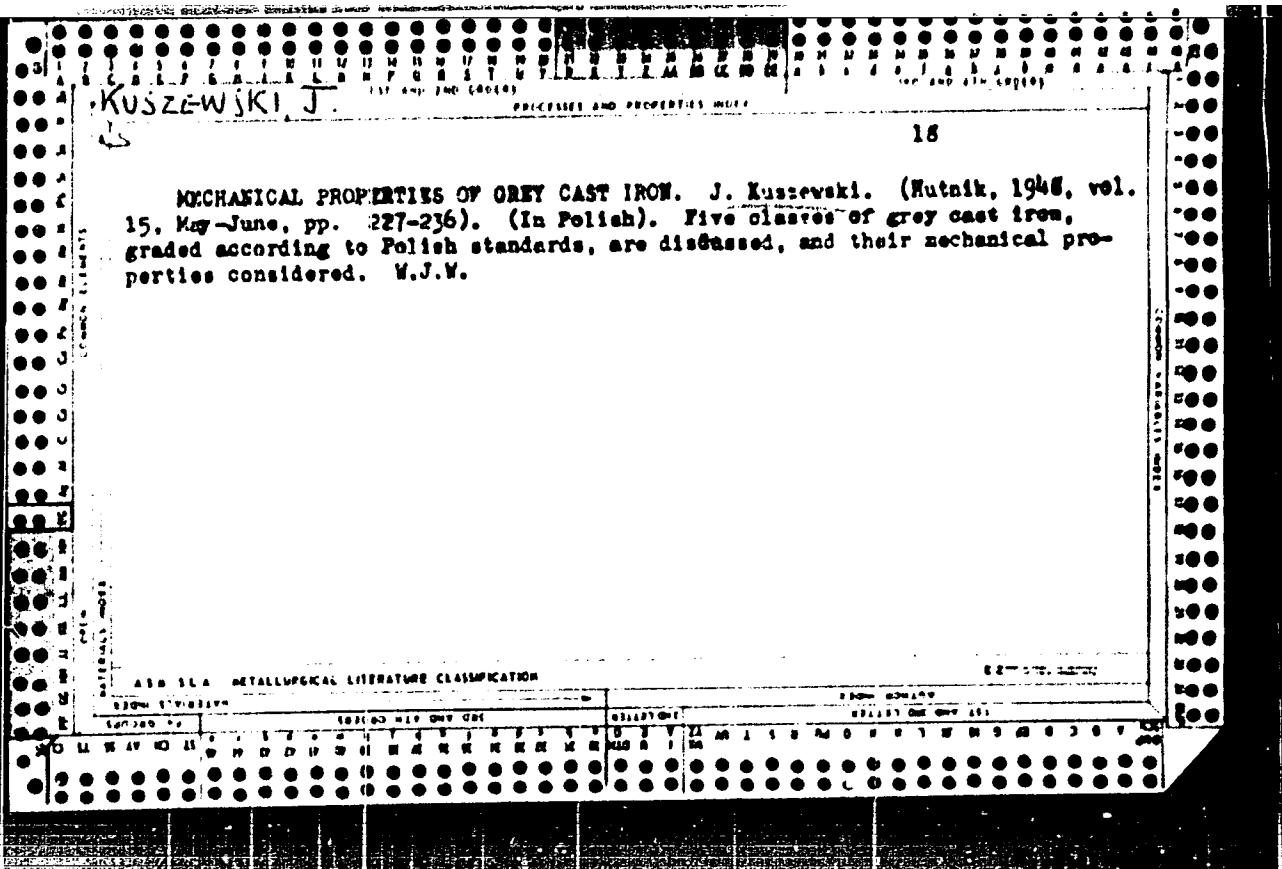
Studies of pulmonary diseases among children of patients in Gruzilka
Gruzilka 32 no.8:695-701 Jan 1984.

I. Z Kliniki Gruzilki Muze Akademii Nauk w Warszawie (bieganowski)
prof. dr. med. J. Kielarowski i z Wydziału Lekarskiego Uniwersytetu
Techniczno-Ekonomicznego w Szczecinie (B. Kowalewski, prof., dr.
h. n. P. Kowalski).

KUSZEWSKI, D. - Rolnik Spoldzielca. Vol 8, no 27, July 1955 - Warszawa

Peace and collaboration p. 1.

SO: Monthly list of East European Accessions List, (SEAL), LC, Vol. 4, No.11
Nov 1955, Uncl.



KUCZEWSKI, J.

"Structure and mechanical properties of cast iron in the light of modern structural diagrams," *Przeglad Odlewnictwa*, Krakow, Vol 4, No 7/8, July/Aug. 1954, p. 205.

SO: *Eastern European Accessions List*, Vol 3, No 11, Nov 1954, L.C.

"APPROVED FOR RELEASE: 03/13/2001

CIA-RDP86-00513R000927910009-4

Korean War

Operations of the Maritime Museum in Tamky, Kwan'gut-maku
File no. 3/4457-453 '64

APPROVED FOR RELEASE: 03/13/2001

CIA-RDP86-00513R000927910009-4"

KUSZEWSKI, Jozef

The Shipbuilding in Poland During the Years 1945-1962
Exhibition arranged in the Maritime Museum in Danzig.
Kwart hist nauki i tech ? nc.3:468-470 '63.

KUZEWSKI, Sces

6607* Casting of Iron Alloys in Permanent Molds. Odlewanie stopów żelaznych w formach trwałych. (Polish) 1971 Kurkowski, Przeglad Poligraficzny, v. 6, no. 2, Feb. 1956, p. 65-69.
Preparation and design of cast-iron molds. Casting techniques and pouring set up, including sprue and gate design. Foundry cost. Cast-iron vs. steel molds, and service life in relation to weight of castings. Diagrams, table.

SJ

LFH

KUSZEWSKI, Zbigniew, inz.

The Oder River during the period 1946-1961. Gosp wodna 22
no.6:243-244 Je '62.

1. Towarzystwo Rozwoju Ziemi Zachodnich, Warszawa.

KUSZTILMAN, K.Sz. [Kusil'man K.S.]

Tasks of the national as well as industrial standardization in
the machine industry. Szabolcs kozl 13 no.5.112-113 My '61.

KUSZKA, Werner; NOWAK, Jan

Obtention of naphtalene of the pressed type by the centrifugation method. Koks smola gaz 6 no.6: 211-214 '61.

1. Zaklady Koksochemiczne, Hajduki

DEKO, Zenon, mgr inż.; KUSZKE, Hubert, mgr inż.

Automatic constant-voltage controlling device for electrostatic precipitators. Energetyka Pol 17 no.11; Suppl.: Energopomiar 9 no.6a37-42 N '63.

1. Pion Elektryczny, Zakład Badań i Pomiarów, Warszawa.

KALI, KLAJUSOW; KUZMIK-JOCHYL, KRISTYNA

N-glycosides of the sodium salt of sulfanilic acid. Roczn
chemii 38 no. 1:17-21 '64.

1. Department of Organic Chemistry, School of Medicine, Krakow,
Division of Pharmacy.

3611

3/081/62/000/009/069/075
B160/B101

15.7140

AUTHORS: Probst, János, Kuszmann, Jánosné, Lipovetz, Iván, Nagy, József

TITLE: Method of making silicone-based heat-resistant and anti-corrosion varnishes

PERIODICAL: Referativnyj zhurnal. Khimiya, no. 9, 1962, 633, abstract
9P327 (Eljárás szilikon alapú hőálló szigetező és
korrozióvédő lakkok előállítására. Hungarian patent 147714,
October 15, 1960)

TEXT: Silicone resin for heat-resistant and anti-corrosion varnishes can be made from polyorgano-siloxane having an R:Si ratio of 1-1.5 and a degree of polycondensation of 10-100, which has been produced from alkyl-, aryl- or alkyl-aryl alkoxysilane by total hydrolysis in the presence of water and p-toluene-sulfo acid (I). The resin is obtained by the interaction of this with a modified polyalkoxy-oligo-organosiloxane having the general formula $\text{RC-SiR}_1\text{R}_2\text{-O}[-\text{SiR}_1\text{R}_2\text{-O}]_n-\text{SiR}_1\text{R}_2\text{-OR}$, where R is an alkyl radical with 1-2 atoms of C whilst R_1 and R_2 are saturated or unsaturated hydrocarbon or

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3/081/62/000/000/269/375
316C/B101

Method of making silicone-based...

Methyl radicals, n = 2-1. (This organosiloxane is obtained by condensation of α,ω -dialkoxy-oligo-dialkyl- or dialkoxy-oligo-alkyl-arylsiloxane with esters of fatty acids containing free hydroxyl groups, 1-5% dicarboxylic acids or their anhydrides being added afterwards to the condensate). Examples: A. Production of silicone resin. (II) 60 g of water are stirred into a mixture of 350 g phenyltriethoxsilane and 1 g I for 3 hours and 1/4
is led for 2 hours. After the alcohols have been driven off, the reaction mixture is dissolved in toluene and any remaining traces of alcohols are removed; the condensation is then continued in a Marcusson apparatus for 3-4 hours while the reaction mixture is boiled in the presence of 3-4 g of zinc stearate. B. Production of silicone plasticizer (III). a) While a mixture of 146 g of dimethyldiethoxysilane with 0.5 g of I is being heated in a water bath for 3 hours, 9 g of water are added; the mixture is kept heated for a further 2 hours and the alcohol driven off. Tetramethyl-diethoxy-disiloxane is obtained. b) A mixture of 44 g of glycerol, 64 g of castor oil and 0.1 g of lead oxide is heated to 250°C in a stream of nitrogen to form a homogeneous mixture. c) The products obtained from (a) and (b) are reacted together and the alcohol driven off in a stream of nitrogen, the temperature being raised from 100 to 200°C in 3 hours. The oily

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method of making silicone-based ...

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reaction mixture so formed is heated to 230° C; 10.6 g of phthalic anhydride are added slowly and the temperature is raised to 240-250° C. After cooling, the product is dissolved in toluene. C. Condensation of II and III is carried out while boiling the toluene. The toluene solution of the varnish is filtered and evaporated to a resin content of 40%. 600-700 ml of a 40% solution of resin is obtained. [Abstracter's note: Complete translation.]

Card 3/3

VARGHA, L.; TOLDY, L.; FEHÉR, O.; HORVATH, T.; KASZTELLNER, L.; KUSZMANN, J.;
LENDVAI, Sarolta

New sugar derivatives with cytostatic effectiveness. Acta physiol.
hung. 19 no.1-4:305-312 '61.

1. Forschungsinstitut fur die pharmazeutische industrie, Budapest.
(CARBOHYDRATES pharmacology)
(ANTINEOPLASTIC AGENTS pharmacology)

KUSZMANN, Karelly; VINCZE, Endre, dr.

Conditions for practical application of clamping devices for
milling machines. Gen 16 no.8:303-311 A6 '64.

1. Borsig Machine Factory, Miskolc (for Kuszmán). 2. Chair of
Mathematics, Technical University of Heavy Industry, Miskolc (for
Vincze).

POLAND / Chemical Technology. Processing of Solid H-22
Fossil Fuels.

Abs Jour: Ref Zhur-Khimiya, No 23, 1958, 78993.

Author : Kuszniariawicz, N. R.

Inst : Not given.

Title : The Experiments in Preparing Shaped Metallurgical Coke on a Pilot Plant Scale in the USSR.

Orig Pub: Koks, smola, gaz, 1957, 2, No 6, 294-299.

Abstract: Methods of preparation, design of a pilot plant unit and experimental results concerning the preparation of metallurgical coke from weakly coking coals are described. The original coal passes through a breaker, a buffer collector, a worm conveyer over for fast heating to a temperature of plasticity, a cyclon for the separation from the vapor phase, buffer capacity, a

Card 1/2

40

APPROVED FOR RELEASE: 03/13/2001 CIA-RDP86-00513R000927910009-
POLAND / Chemical Technology. Processing of Solid H-22
Fossil Fuels.

Abs Jour: Ref Zhur-Khimiya, No 23, 1958, 78993.

Abstract: compressor and an oven for the thermal treatment of the briquets obtained. Data is furnished on physical-mechanical properties of the prepared coke as well as data on the characteristic of the resulting liquid and gaseous products.

Card 2/2

KUSZTOS, Denes, dr.; KELLER, Laszlo, dr.

Rheography. Orv.hetil. 102 no.6:251-254 5 F'61.

1. Fovarosi Istvan Korhas, I. Belosztaly.

(HEART physiol)

(BRAIN physiol)

KUSZTOS, Denes, dr.; SARKOZY, Katalin, dr.; VARNAI, Gyorgy, dr.

Amputation of patients with vascular diseases with special
reference to surgical risks. Orv.hetil. 102 no.10:454-456
5 Mr '61.

1. Fovarosi Istvan Korhaz, I. Belosztaly.
(AMPUTATION)
(VASCULAR DISEASES PERIPHERAL surg)

KELLER, Laszlo, dr.; KUSZTOS, Denes, dr.

Regression in diabetes mellitus caused by cerebrovascular insultus.
Orv. hetil. 102 no.19:896-898 7 My '61.

1. Fovarosi Istvan Korhaz, I Belosztaly, Budapest.

(DIABETES MELLITUS) (CEREBRAL HEMORRHAGE)

BODROGI, Gy.; KALMAN, P.; KUSZTOS, D.

On the role of rheocardiography in the determination of single heart
phases. Acta med. Hung. 18 no.2:189-196 '62.

1. Staatliches Kardiologisches Institut, IV. Medizinische Klinik
der Medizinischen Universitat, und I. Innere Abteilung des Istvan-
Krankenhauses, Budapest.
(ELECTROCARDIOGRAPHY)

SZEPLAKI, Sandor, dr.; KUSZTOS, Denes, dr.

Data on the occurrence and pathomechanisms of cor pulmonale and
chronic bronchopulmonary disorders according to clinical examination.
Orv. hetil. 103 no.29:1349-1354 22 Jl '62.

1. Fovarosi V. ker. Szkarvosi Rendelo (Rosenberg hp. u.) EKG. es
Istvan Korhaz, I. Belosztaly.
(PULMONARY HEART DISEASE statist) (LUNG DISEASES statist)

KUSZTOS, Denes, dr.; KELLER, Laszlo, dr. FANO, Jozsef, dr.

Rheographic examination of the effect of syncardial massage.
Orv. hetil. 105 no.12:550-553 22 Mr'64.

1. Fovarosi Istvan Korhaz, I.Belosztaly.

*

KUSZTOS, Denes, dr.; HAFFNER, Zsolt, dr.

Use of Peripherin-Homburg in circulatory diseases of the brain.
Orv. hetil. 105 no.21:991-993 24 My'64

1. Istvan Korhaz, I. Beloaszaly es Idegosztaaly.

*

BUGAR-MEZSAROS, Károly, dr.; KUSZTOS, Dezső, dr.

Our therapeutic results with the Syncardon machine. Orv. hetil.
105 no.52;2467-2470 27 D '64.

1. Föveros! Istvan Korhaz I. Belosztaly (foorvos: Bugar-
Mezsařos Karly dr.).

BIEK, Jaros, diplomatic representative; BAKI, Jozef, engineer; BIEK, Gyorgy; KOMACZ, Lajos; MEGYERY, Gyorgy; MOLNAR, Jozsef; TOTH, Jeno; VENCZAK, Vilmos

Efficiency of gas consumption of various consumers with regard to the optimum distribution of energy carriers. Proceedings atom 18 no.1:15-17 Ja '65.

1. National Petroleum and Gas Industry Trust, Budapest [for Pora], & Ministry of Heavy Industry, Budapest [for Venczak]

KUSZYNSKI, Wlodek; GLEWICZ, Janusz

Swelling kinetics of brown coal and peat in pyridine. Mat
chemia no.6:26-33 '62.

1. Katedra Technologii Chemicznej, Uniwersytet im. Adama
Michniowicza, Poznan.

KUT, Katei

Testing the sensitivity of the AECI to the parameter
of the revision of the initial value of the initial reac. heat flux.
Report 1C Rev. 305 D 364.

1. Institute of Gas Technology and Research, Inc., Chicago.

KUTA, A.

Occupational factor in cutaneous cancer. Bratisl. lek. listy 32
no. 3-4:189-192 1952. (CLML 24:1)

1. Of the First Dermatological Clinic of Charles University, Prague.

BLAZEK,J., MUDr; CERNY, E., MUDr; KUTA, Adolf, MUDr, asistenti kliniky

Porokeratosis of Mibelli. Cesk.derm. 31 no.4:185-189 Aug 56.

1. Z I. dermatov. kliniky KU (predn. prof. MUDr Karel Gawalowski)
(KERATOSIS, case reports
porokeratosis of Mibelli (Cz))

EXCERPTA MEDICA Sec 13 Vol 13/6 Dermatology June 59

1554. TISSUE THERAPY IN SCLERODERMA. THERAPEUTIC EXPERIENCES AND A SUGGESTION FOR A METHOD OF APPLICATION - Tkáčová
therapie chránícené sklerodermie - Kdta Ad. I. Dermatol. Clin.,
Charles Univ., Prague - ČSL.DERM. 1957, 32/2 (72-75) Illus. 6

Results of tissue therapy in 44 cases of scleroderma of various clinical types treated since 1948 are evaluated. Whereas in other skin diseases tissue therapy does not exceed or even reach the results of other treatments, tissue therapy is most successful in both the solitary (scleroderma 'en bandes') and the disseminated form of scleroderma. From 15 patients treated for solitary or disseminated scleroderma, 11 were cured and 4 interrupted treatment. Splendid results were attained with this method in cases of scleroderma 'en bandes'. Six cases of superficial scleroderma of the type of Unna's 'Kartenblattähnliche Sklerodermie' were also successfully treated. Owing to the necessity of prolonged treatment, the author modified the original implanting method as follows: the preserved tissue of amniomembrane (sterilized before use) is reduced to a pulp with the scalpel, then injected with a thick needle under the changed skin. These injections are repeated at 14-day intervals and usually after 6-10 injections the first obvious softening appears. Only exceptionally does inflammation appear. This therapy can be quite safely recommended.

A.L.H.A.
EXCERPTA MEDICA Sec.13 Vol.12/4 Derma/Venereology Apr58

695. KOEBNER'S PHENOMENON IN A STUDY CONCERNING THE PRIMARY EPIDERMAL PATHOGENESIS OF PSORIASIS - Kútá A. and Neumann E. Dermato-Venereol. Clin. of Charles Univ., Prague - DERMATOLOGICA (Basel) 1957, 115/1 (51-60) Illus. 6

The authors support the view that, in the pathogenesis of psoriasis, an increased regenerative activity of the epidermis cells is the primary and determining process. To prove this conception, they have tried to demonstrate such processes in the epidermis at a time when no changes can be seen in the corium. They have used for this purpose, on the principle of Koebner's phenomenon, the mechanical stimulus of a scratch on a piece of skin, then excised the piece before the appearance of any manifest symptoms, and tested it for dehydrogenase activity, which is known in general to accompany proliferative processes. This activity was in fact seen to be significantly increased in the epidermis of psoriatic patients compared with that of control persons.

ALGERIA MEDICA Sec. 13 Vol. 11/9 Dermatology Sept. 57

2009. KUTA A. I. Dermatol. Klin., Karls Univ., Prag. "Pellagra acuta. Acute pellagra DERMATOL. WSCHR. 1957, 135/2 (25-28) Illus. 2 Eight cases of acute pellagra, which were observed in North Korea are briefly described. Because of the highly insufficient nutrition, especially in respect of proteins, minerals and vitamin (niacin), the conclusion is reached that the conditions for the development of pellagrous deficiency photosensitization were present. The acuity of the development and of the phenomena was striking. The quick result of vitamin therapy also suggests acute pellagra, in which gastrointestinal disturbances and symptoms are either of secondary importance or absent.

(XIII, 6*)

EXCERPTA MEDICA Sec 13 Vol 13/5 Dermatology May 59

1140. ROLE OF THE SKIN ADNEXAE IN THE PATHOGENESIS OF PSORIASIS - Neumann E. and Kuta A. First Dermato-Venereol. Clin., Charles Univ., Prague - DERMATOLOGICA (Basel) 1958, 116/6 (460-468) illus. 5

A constant, strongly positive dehydrogenase reaction in the skin adnexae, and particularly in the hair follicles and their ostia opening into the epidermis, suggests a high regenerative and functional activity in this apparatus. In pursuance of their previous findings, that psoriasis is a primary epidermal hyperregenerative process, and that it occurs chiefly in regions of a physiologically high regeneration of the epidermal cells, the authors histologically studied initial lesions in different clinical forms of psoriasis, with special regard to the follicular apparatus and to the adnexae in general. Microscopy conclusively showed that the psoriatic process in the epidermis develops from the ostia of the follicles and possibly also of the sweat glands, whereas changes of a simple hyperregenerative nature occur even in the hair follicle proper. In the light of this finding, the authors attempt to explain a series of clinical features of the psoriatic process.

KUTA, Adolf

Mibelli's porokeratosis zoniformis. Report of 3 cases and etiopatho-
genic analysis. Cesk. derm. 34 no.2/3:122-130 Ap '59.

l. I dermatovenerologicka klinika KU, prednosta prof. dr. K. Gawalowski.

(SKIN dis)

KUTA, A.

Lichen ruber plano-pilaris decalvans, resp. debarbans. Contribution
to its nosology and therapy and considerations on pseudopelade.
Sborn.lek.62 no.11:315-330 N°60.

1. I. dermatologicka klinika fakulty všeobecného lekarství Karlovy
University v Praze, prednosta prof.dr. K. Gavalowski.
(LICHEN PLANUS compl)
(ALOPECIA etiol)

KUTA, A.

Carcinomatosis cutis teleangiectatica secundaria. Sborn.lek.62
no.11:324-326 N°60.

1. I. dermatovenerologicka klinika fakulty všeobecného lekarství
university Karlovy v Praze, prednosta prof.dr. K.Gawalowski.
(SKIN NEOPLASMS case reports)
(CARCINOMA case reports)

KUTA, Adolf

Peroral antidiabetic Tobucin-SPOPA in the treatment of juvenile
verruca plana. Cesk. derm. '66 no.6:396-400 '61.

1. I dermatovenerologicka klinika University Karlovy v Praze, pred-
nosta prof. dr. J. Konopik, Dr. Sc.

(PAPILLOMA ther) (ANTIDIABETICS ther)

ERBAK, V.; JILEK, M.; KUTA, Ad.

2 cases of Mibelli's porokeratosis (Case report contribution with an etiopathogenetic analysis). Cesk. derm. 36 no.1:51-54 F '62.

1. Dermato-venerologicke oddelenie OUNZ v Rimavskej Sobote, prednosta prim. dr. V. Erbak I. dermatovenerologicka klinika University Karlovy v Praze, prednosta prof. dr. K. Gawalowski.
(SKIN diseases)

KUTA, A.

Our experience with surgical therapy of rhinophyma. Cesk. derm. 37
no.4:276-277 Ag '62.

1. I. dermatovenerologicka klinika fakulty všeobecného lekarství
University Karlovy, prednosta prof. dr. J. Konopík, DrSc.
(ROSACEA surg)

KUTA, Adolf

A contribution to the problem of malignant melanoblastoma.
Sborn. lek. 44 no.2:46-56 F '62.

1. I dermatovenerologicka klinika fakulty všeobecného lekařství
University Karlovy v Praze, prednosta prof. DrSc. MUDr. J. Konopík,
(MELANOMA pathol.)

KUTA, Adolf

Therapeutic problems in cutaneous carcinoma from dermatologist's viewpoint. I. Cas. lek. cesk. 101 no.43:1273-1279 26 0 '62.

1. I. dermatovenerologicka klinika fakulty vseobecneho lekarstvi KU
v Praze, prednosta prof. dr. J. Konopik.
(SKIN NEOPLASMS)

KUTA, A.

Chronic roentgen dermatoses with multiple carcinomatous growths and
skin sarcoma. Cesk. rentgen 17 no.2:116-123 Mr '63.

l. I. dermato-venerologicka klinika fakulty vseobecneho lekarstvi
KU v Praze, prednosta prof. dr. J. Konopik, DrSc.
(RADIATION INJURY) (SKIN NEOPLASMS) (CARCINOMA EPIDERMOID)
(CARCINOMA BASAL CELL) (SARCOMA)

KUTA, A.

Chronic late roentgen dermatosis with cancerization after
application of very soft (Bucky) roentgen rays. Cesk.
rentgen. 17 no.4:273-279 J1 '63.

1. I dermato-venerologicka klinika fakulty vseobecneho
lekarstvi KU v Praze, prednosta prof. dr. J. Konopik, DrSc.
(RADIATION INJURY) (DERMATITIS)
(SKIN NEOPLASMS) (ACNE) (CARCINOMA, BASAL CELL)

JILEK,M.; KUTA,A.

Werner's syndrome. Clinical comparisons in histological picture of skin changes. Sborn. lek. 66 no.2:55-60 F'64.

1. I.dermato-venerologicka klinika fakulty vseobecneho lekarstvi University Karlovy v Praze; prednosta: prof.dr. J.Konopik, DrSc.

*

KUTA, A.

Etiopathogenesis of psoriasis vulgaris in the light of its
simple manifestations. Cesk. derm. 29 no.3 193-199 My'64

I. I. dermatovenereologicka klinika fakulty vseobecneho
lekarsv KU [Karlov university] v Praze; prednosta: prof.
dr. J. Konopik, DrSc.

KITA, A.

Therapeutic problems in skin carcinoma from the viewpoint of
the dermatologist. Onk. Ber. Cook. 103 no. 17-47-473 ap. 24. 1964.

L. T. dermatovenerologické klinice fakultního výzkumu Československé
A.U [Fakulta Univerzity v Praze] (prof. Dr. M. J. Šimánek,
DrSc.).

KODAK, Inc.

It is, however, to be noted that the present system of classification is not entirely satisfactory.

1. Katedra dermatologii i lekarstw UW w Krakowie, Wydział Lekarski Uniwersyteckiego Związku:
dermatologo-venereologiczny, ul. Warszawska 11, tel. m. 3848
w sprawie S. na Podkarpackim, tel. m. 3848, 3849.

KUTA, A.

Some remarks on further educational aims of the department
of dermatology and venerology of the Medical Institute for
Graduate Studies. Cesk. derm. 40 no.6; 1968-411 - 9 '65.

1. Katedra dermatovo-venerologie UML v Praze (vedouci doc. dr.
A. Kuta, CSc.).

KUHA, F.

Relation between the production of agricultural machinery and agriculture from the point of view of research.

p. 43 (Zemedelske Stroje) Vol 2, no 2, Feb. 1957 Brno, Czechoslovakia

SO: Monthly Index of East European Accessions (MIEA) LC, Vol. 7, No. 1, Jan. 1956

KUTA, I.; EISELT, L., MUDr.; BOJANOVSKY, I.; BOSAK, V.

Sport efficiency and strength in the aged. Cas. lek. cesk. 104
no.13:351-356 2 Ap '65

1. Vyzkumný ústav telovýchovy v Praze (ředitel: MUDr. E. Eiselt).

KUTA, IA.; SMOLER, I.

Instantaneous polarographic currents. III. i-t curves of the diffusion current with a high concentration of amalgam-forming depolarizers.
Coll Gz chem 26 no.1:224-229 Je '61. (EEAI 10:9)

1. Poliarograficheskiy institut, Chekhoslovatskaya Akademija nauk,
Praga.

(Polarograph and polarography) (Diffusion)
(Depolarizers)

CZECHOSLOVAKIA

BRANICA, M; KUTA, J

1. Department of Physical Chemistry, "Ruder Boskovic" Institute, Zagreb, Yugoslavia - (for ?) 2: J.Heyrovsky Institute of Polarography, Czechoslovak Academy of Sciences, Prague - (for ?)

Prague, Collection of Czechoslovak Chemical Communications.
No 7, July 1966, pp 2835-2840

"Polarographic study of reduction and dismutation of uranium in aqueous solutions of acetylacetone."

CA

Hydrogen overvoltage at the dropping-mercury electrode with constant drop time. Kato (Central Products Lab., Institute for Physical & Chemical Investigation, Stuttgart, Germany) has reported the H₂ overvoltage, characterized by its half-wave potential $E_{1/2}$, as reported in strong acid. For 10% NaClO₄ in 1 N HCl, $E_{1/2}$ is 41 mV more negative than in 0.1 N NaClO₄. These results are not changed by the presence of Na⁺, Cl⁻, Cu²⁺, Cr²⁺, or Ti⁴⁺. At short electrolysis times, and Br⁻ (0.1 to 1.0 M) effect $E_{1/2}$ equally, i.e., $E_{1/2}$ is more negative and in NaClO_4 21 mV more neg. than in 0.1 N HCl. Increasing the height of the Hg reservoir displaces $E_{1/2}$ by 10 to 15 mV. In 0.1 N KCl, KBr, KSCN, or BaCl₂, about 24 mV. more neg. The plot (square root of Hg height vs. wave height) shows that the H₂ reduction is diffusion-controlled. A special app. permitting artificial control of stop time, i.e., the effects of t and m , the viscosity of solvent. With t const., $E_{1/2}$ is independent of m . In absence of t (from 0.4 to 4.3 sec.) shows that H₂ dependence is irreversibly $E_{1/2}$ depending upon t (daher, $E_{1/2} = R/T/Ft$). In t , wave height deviations from the linear equation, particularly at small t , are attributed to much removal of Hg drops. In 0.1 N inert electrolyte, $E_{1/2}$ is independent of H₂ concn. (whether from HCl or H₂SO₄). The contradictory results of Tomita (C.A. 31, 2579) are attributed to wave deformation resulting from maxima at onset of H₂ reduction, phenomena, and (also) blue effect $E_{1/2}$ for 10% N HCl. A derivation is given for the formula $E_{1/2} = E_{1/2} + R/T/Ft$ in $(\text{Hg} - n)^{1/2} + \text{const}$ for the case with equal results that $E_{1/2}$ is independent of n and of the H₂ concn. The value ($\Delta E_{1/2} = R/T/Ft$) in 10% HCl is 17.5 mV. I check the const. shift of 16 mV caused by doubling the height of the Hg reservoir. J. P. Haslett

24(2.4)

VII

CZECH/2332

VI. 1. POOL INFORMATION
CZECH/2332

International Polarographic Congress. Int. Prague, 1951
 Internationale Polarelektrolytische Konferenz. Int. Prag, 1951
 Spomín. I. Mezinárodní polarelektrického sjezdu. Olomouc, 1951. [Miami
 Conference na polarelektrolytice. Význam. Průběžná výstava.]
 Reference researches on polarography in concentrated sulfuric acid
 held at the Congress. [Prague, 1951]. Průběžná výstava
 held at the Congress. [Prague, 1951]. Publishing House
 Tři P. 2,000 copies printed.

Resp. M. J. Koryta, Doctor; Chief Ed. of Publishing House

Milan Stehlík, Doctor; Tech. Ed.; Editor-Director,
 and physiologist.

Rumpál, The book is intended for chemists, chemical engineers,
 and physiologists. The book is a collection of reviews and original papers
 concerning the book is a collection of reviews and original papers
 read at the International Polarographic Congress held in Prague
 read at the International Polarographic Congress held in Prague, which
 read at the International Polarographic Congress held in Prague, which
 In 1951. Uses of polarography in electrochemistry, industrial
 biochemistry, medicine, Read at the Congress. Russian and English which
 either German or English. Original papers, and English which
 presented. In the section in Russian, German, and English which
 only those translations in Volume 1 are represented. The
 have not been published in Volume 1 are represented. The
 following scientists participated in the organization of the Congress:
 Comessi, Vanni, Doctor; Jakovljević, Dr., Professor of
 Sciences; Professor Jaroslav Herčík, Chairman of the
 Plenary; Professor Jaroslav Herčík, Chairman of the
 of the Congress; and Scientific Research and Technical
 Development. References follow each paper.

- Forejt, J. - Preparation for Oscillographic Polarography 241
 [Russian Translation] 250
 [German Translation] 259
- Hrboček, J. - Oscillographic Polarography 269
 [Russian Translation] 270
 [English Translation] 273
- Pudlinský, R. Kinetics of Electrode Processes in Polarography 296
- Card 1/14

- Arend, H. T. Polarographic Study of Basic Trivalent
 Chromium Salt Systems 395
- Krivanek, M. Complexes of Iron with Saccharose 399
- Drotorsky, M. and M. Drotor. Effect of Gelatin and Phenoxy
 on Electrode Deposition of Ions at a Dropping Mercury
 Electrode 404
- [Russian Translation] 407
- [German Translation] 410
- Fulštejnský, J. Study of Hydrogen Overvoltage with a Mercury
 Electrode With Controlled Dripping Time 413
- Dvořák, J. Effect of Capillary Constants on the Wallman
 of Oxygen [Russian Translation] 418
- [German Translation] 421
- Vavříček, J. Attempt to Classify Refined Sugars by the
 Polarographic Method [Russian Translation] 427
- [German Translation] 432

ORIGINAL PAPERS READ AT THE CONGRESS

- Kaloušek, M. and A. Pockstein. Validity of the Nernst
 Equation in the Description of the Polarographic Wave Edge 359
- Vlček, A. A. Polarography in Concentrated Sulfuric Acid 366
- [Russian Translation] 370
- [English Translation] 373
- Valente, P. Study of Current Discontinuity Appearing on
 a Cyclic Polar Electrode 377
- Kasick, J. Discontinuity on Polarographic Curves Observed 381

KUTA, J., RIMA, J., NOWAK, J.

"Contribution to the Polarographic Determination of Manganese and Iron" p. 49,
(CHEMICKE LISTY, Vol. 47, no. 5, May 1953, Praha, Czechoslovakia).

SO: Monthly List of East European Accessions, LC, Vol. 2, No. 11, Nov. 1953, Uncl.

[Handwritten note: 1/20/64]
The mechanism of hydrogen liberation at mercury electrodes. Jorgová, E., T. C. Skoček, and J. Prugovec. *Chem. Zvesti* 1963, No. 4, p. 353-354. The mechanism of the liberation of H₂ caused by the overvoltage of the static and dropping Hg electrodes was studied. The components

of the overvoltage of H⁺ were investigated and measurable on Hg electrodes, especially by the dropping Hg method which was essentially a polarographic method (1). New overvoltage laws such as the derivation of the logarithmic dependence of overvoltage on the Hg drop time and concn. of acid applied. Polarographic criterions for limiting currents showed by what processes the current was limited. In the case of fast reactions the current was limited by diffusion, but at certain potential it was proportional to $t^{1/2}$ (t = Hg drop time) which corresponded to the diffusion current. If the current was limited by the rate of the reaction, it was proportional to $t^{1/2}$ and agreed with the kinetic current. The limiting currents of the weak and irreducible acids behaved as diffusion currents. They decreased after the anions of these acids were added, and simultaneously the half-wave potentials shifted in a neg. direction. In the case of H₂BO₃, the current was of a kinetic nature. A possibility was the detn. from the polarographic data, of the consts. of dissociation and recombination of weak irreducible acids. 50 references. Jan Michal

KUTA, JAROSLAV

Polarographic behavior of boric acid. [Janice E. Haid (Institute of Inorganic Chemistry, Prague). Chem. Zentralblatt (1954).]—In 0.1M (CH₃CN) soln, a wave was observed at -1.07 v. against a normal Hg/Cl electrode which at 25°C. time is smaller than corresponded to the diffusion coefficient of reduction of the acid. There was a kinetic control of the temp. coeff. of which was only 2.3% per degree. On addition of polyalcohols, e.g. of glycerol or sorbitol, the wave height increased and the half-wave potential was shifted to more pos. values. With excess of glycerol, the current remained kinetic and had the same temp. dependence, while with excess of sorbitol the current became diffusional. The wave was ascribed to the reduction of H₃O⁺, which was formed by a relatively slow dissociation, which preceded the electrode process. E. Kralov.

KUTA, JAROSLAV

CZECH

Catalytic currents due to pyridine in unbuffered solutions.
Jaroslav Kuta and Josef Jirsík. Československý chemický list 52(1),
July 49, 21-27 (1955). In solns. of pyridine (I) concn. 0.1N KCl and 1N HCl or a weak non-reducible acid of $pK = 2.5$, 2 waves were found at the potentials -1.53 and 1.8 v. (vs. Hg calomel electrode). The more pos. wave was of diffusional nature and was due to the reduction of H^+ . In the presence of I this wave was shifted towards more pos. potentials and its slope changes were due to the catalytic action of I. The height of the more neg. wave increased with decreasing height of the Hg column and with increasing concn. of the acid. With decreasing concn. of I it tended toward a limiting value. This wave was due to the catalytic reduction of H^+ from H_2O and from the diffusion current as Pošekova and Geplová (Č.Č., 47, 8125c) concluded. In the solns. of 0.1N LiCl and LiCl this only wave was observed at -1.8 v. Its height decreased with increasing concn. of 10% I, and in 50% PDD it was independent of the height of the Hg column.

E. Štrádová

KUTA, J.

CZECHOSLOVAKIA

No academic degree indicated

Polarographic Institute, Czechoslovak Academy of Sciences (Polarographisches Institut, Tschechoslowakische Akademie der Wissenschaften), Prague

Prague, Collection of Czechoslovak Chemical Communications, vol 27, No 10,
Oct 62, pp 2349-2364.

"Influence of Surface-Active Substances on Polarographic Currents VI. Momentary
Currents and Gradated Formations in Some Reversible Systems in the Presence of
Rapidly Absorbed Charged and Uncharged Surface-Active Substances."

Co-author:

SMOLER, I., Polarographic Institute, Czechoslovak Academy of Sciences (Polaro-
graphisches Institut, Tschechoslowakische Akademie der Wissenschaften), Prague

KUTA, J
CZECH

Polarographic reduction of oxalic acid. J. Kuta (Praha-
řežský ústav ČSAV, Prague). Chem. Listy 47: 261-2
(1953). A measurable polarographic wave of a height

directly proportional to the concn. of oxalic acid can be
obtained in solns. contg. 3×10^{-3} N HCl and 1-3N KCl,
the concn. of oxalic acid not being lower than 1×10^{-4} N.
The half-wave potential is -1.0 v. P. Strafelda

AT 54

K. J. A. R. D. S. C. P. 10
✓ The rate of recombination of oxalic acid on the dropping mercury electrode. Jaroslav Kola (Polarografický ústav ČSAV, Prague). *Chemistry* 1975, 14(7-75) (1968).—The 2-electron reduction wave of oxalic acid (1) had a diffusive nature and decreased between pH 3 and 6, in accordance with the dissociation curve of I ($pK = 4.48$). The undissociated, I, which was subject to reduction, was supposed as being formed by the recombination of its ions near the electrode. By using the equations of Koutecký (C.I., 48, 3051g), the first rate const. of the recombination of I was $k_r \sim 5.5 \times 10^{-4}$ l./mol. sec. (0.2M Britton-Robinson buffer, solns.) (2.7N KCl). The pH value in the neighbourhood of the electrode remained const., when the concn. ratio of I and the buffer soln. had a value 1:20.

KUTA, JAROSLAV

✓ The effect of electrolyte solutions on the reduction of
oxalic acid. / Jaroslav Kuta (Czechoslov. Acad. Sci.,
Prague). *Z. f. Chem. Acad. Sci. Hung.* 9, 119-52 (1953).

(in German) (English summary); cf. *C.A.* 51, 1185c. --
The polarographic reduction of oxalic acid in phosphate or
citrate buffers (pH 2 to 3), or in 0.01-0.1M HCl, is not
observable before the H^+ ion reduces, but if the soln. is
made *N* in KCl, a wave appears whose height is propor-
tional to the oxalic acid concn. A large KCl concn. (ap-
prox. 2M) leads to a max. of the 2nd kind. $MgCl_2$ (0.1 to
1M) or sulfate lowers the effect; NH_4Cl or KI act
similarly to KCl. Increasing the pH lowers the wave
height, and this effect is increased at the lower temps. The
KCl effect is accompanied by a shift of the oxalic acid reduc-
tion potential to more pos. voltages. The half-wave poten-
tial depends on the amt. of KCl used. The influence of
the nature of the supporting electrolyte on the polarographic
behaviors of a number of types of org. substances is also re-
viewed. *by J. L. Johnson*. *H. K. Zingg* [initials]

~~Kutná Hora, Czechoslovakia~~

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*✓ Determination of the rate constant of the dissociation of
boric acid from polarographic limiting currents. Jaroslav
Kára (Polarografický ústav ČSAV, Prague), ČAS, Tří-
leté práce v USA. "Polarographic kinetic currents of boric
acid (I) are a function of the rate of dissociation. The curve of
current vs. current of I obtained on the dropping-Hg electrode
in the case of neutral salts is not linear but is convex to
ward the current axis. By the method of least squares proposed by
Hansen (C.J., 50, 70c), the rate constant of the dissociation of I
was determined as $k_1 = 1.3 \times 10^{-4}$ sec.⁻¹. [Signature]*



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KOTF, TARASLOV

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[Signature] Hydrogen overpotential [Signature] C. R. Argon
Prague, Czech. Rep. [Signature] Reviewed
of interelectrolytic bases and general cathodic and anodic
processes.

[Signature]

APPROVED FOR RELEASE: 03/13/2001

CIA-RDP86-00513R000927910009-4"

Bata, J.

"Determination of the dissociation speed constant of boric acid from the polarographic limiting currents." In German."

p. 141' (Collection of Czechoslovak Chemical Communications. Vol. 27, no. 5, Oct. 1957, Praha, Czechoslovakia.)

Fentaky Index of East European Acquisitions (EEA) 1C, Vol. 7, no. 7, July 1958

HUFA, J.

"Effect of the basic electrolyte on the reduction of oxalic acid ester
in formic acid."

p. 1677 (Collection of Dr. Čeněk Václav Chemical Communications, Vol. 10, No. 5,
Oct. 1957, Praha, Czechoslovakia.)

Monthly Index of East European Accessions (MIEA) 10, Vol. 1, No. 1, July 1956

Kuta Jaroslav

CZECHOSLOVAKIA/Physical Chemistry - Electrochemistry.

B-12

Abs Jour : Ref Zhur - Khimiya, No 8, 1958, 24330

Author : Kuta Jaroslav

Inst :

Title : Effect of Extraneous Electrolyte on Reduction of Esters
of Oxalic Acid.

Orig Pub : Chem. listy, 1957, 51, No 4, 764-766

Abstract : In continuation of prior researches (RZhKhim, 1955, 42700; 1956, 71313; 1957, 40768) it is shown that on polarographic reduction of diethyl ester, monoethyl ester and their anions, and also of anion of oxalic acid (I) there is observed a shifting of their waves in the positive direction in the presence of polyvalent cathions (Al^{3+} , La^{3+} , Ca^{2+} , Ba^{2+}). In unbuffered solutions these cathions increase the height of the wave. Hydrolysis of diethyl ester of I is catalyzed by bases.

Card 1/1 Polarographic Institute
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CZECHOSLOVAKIA / Physical Chemistry Electrochemistry. B

Abs Jour: Ref Zhur-Khimija, No 11, 1958 35 556

Author : Kuta Jaroslav

Inst : Not given

Title : Comparison of Hydrogen Excess on Dropping and
Flowing Mercury Electrodes.

Orig Pub: Chem. Listy, 1957, 51, No 7, 1274-1282

Abstract: Polarographic curves have been plotted with the application of dropping and flowing Mg-Electrodes, at relatively high concentrations (up to 0.1 N) of strong acids in the presence of an indifferent electrolyte overtension at $1.3 \cdot 10^{-5} - 0.1$ a/cm⁻². A timing device for the regulation of the drop formation has been applied in the case of the dropping electrode. The (i,t) curves of the first drop have been plotted. The experimental

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CZECHOSLOVAKIA / Physical Chemistry. Electrochemistry. B

Abs Jour: Ref Zhur-Khimiya, No 11, 1958, 35 556

Abstract: findings for the mean current density value i on the dropping electrode are expressed by the equation $E = \text{const} + (2RT/F) \ln[H^+] - (2RT/F) \ln i$. The time dependence of the instantaneous current ($i = kt^n$) varies with the applied voltage before reaching the limit current i_d , so that n drops from 0.6 (lower section of the curve) to 0.5 (at $E = 1/2$) and to 0.22 (after reaching i_d). On the application of a flowing electrode i_d is determined by a diffusion. In this case $E = 1/2$ is negative in relation to $E = 1/2$ of a dropping electrode with respect to the value $(RT/F) \ln(t/t_1)$ (t - contact time of Hg-flow with the solution; t_1 -- dropping period). Equations permitting to compare over-

Card 2/3

CZECHOSLOVAKIA / Physical Chemistry. Electrochemistry. B

Abs Jour: Ref Zhur-Khimiya, No 11, 1958. 35556

Abstract: tension measurements at 1 constant with polaro-graphic overtension measurement have been derived.

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